

# NUMERICAL PROBLEMES

1. A bus travel 15km toward west and makes u-turn back to travel a further distance of 10km find.

a) Distance travelled      b) it displacement?

Distance covered west – ward = 15km

∴ ∴ ∴ East – ward = 10km

According to given figure.

Distance travelled = AB + BC = 15 + 10 = 25 km.

Displacement = AB – BC = 15 – 10 = 5km.



2. A race car travel around a circular track, covering a distance of 580m in 25 mints before stopping at point from where it started determine average velocity of the car during this period?

Cover distance = 850m time =  $t = 25\text{sec}$  average speed =  $\langle v \rangle = ?$

$\langle v \rangle = \text{total displacement} / \text{total taken time}$  as the body moves on a circular path and comes back to its starting point, hence displacement is zero as shown.

$\langle v \rangle = 0 / 25 \Rightarrow \langle v \rangle = 0$ .



3. A truck moving at a speed of 20m/sec begins to slow at constant rate of  $3\text{m/sec}^2$ . Find for it goes before stopping?

Initial speed =  $v_i = 20\text{m/sec}$

Initial speed =  $v_i = 20\text{m/sec}$

Final speed =  $v_f = 0\text{m/sec}$

Acceleration =  $a = -3\text{ m/sec}$

Distance covered before stop's = ?

$2as = v_f^2 - v_i^2$
$2 \times -3 \times s = 0^2 - 20^2$
$-6s = -400$
$-6s = \frac{-400}{-6} = s = 67\text{m}$

4. The speed of a bus reducing uniformly from 15km/sec to 7m/sec while travelling a distance of 90 m (a) find acceleration (b) further covered distance before stopping

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**CL Gary**



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**Chemical Thermodynamics** Victor CM Freestone,2025-06-13 Thermodynamics can never be made easy but with the right approach and a consistent use of scientific terms it can be made less opaque and it can give a person who is prepared to try an insight into how science explains why things happen the way they do The approach adopted in this book will give readers a better understanding of how science works together with its limitations Unfortunately thermodynamics or at least some parts of it is a subject which apart from quantum mechanics probably causes most confusion and bewilderment amongst scientists The majority of students do not understand or get thermodynamics and it is considered a hard or difficult subject There are multiple reasons for this There is of course mathematics and many thermodynamic texts appear to be lists upon lists of differential equations Another reason is that thermodynamics is as often as not poorly taught by teachers lecturers who themselves do not understand or appreciate or have any interest in the subject often all three This results not only in a lack of scientific rigorousness in the teaching of the subject with the resulting confusion and sometimes teachers lecturers and authors just get it plain wrong this occurs surprisingly often However it need not be like this and although mathematics including calculus is required it can be kept to a relatively elementary level in order to obtain an understanding of this most important of subjects No one can pretend that the subject is easy but it can be made more accessible by a rigorous definition of terms and concepts and ensuring that a consistency of use of these definitions is maintained Highlighting the benefits of thermodynamics in practical science the text gives an intuitive grasp of the major concepts of thermodynamics such as energy and entropy Provides a new pedagogic approach to understanding and teaching chemical thermodynamics Starting with a set of basic simple assumptions about what constitutes topics such as an ideal gas theories are developed in a clear concise and accessible manner that will either answer or at the very least give an insight into a surprising range of scientific phenomena including energy heat temperature properties of gases time and quantum theory Assumes that the reader has essentially no knowledge of the subject Mathematics including calculus is kept to a relatively elementary level in order to obtain an understanding of this most important of subjects Provides the reader with a better understanding of how science works together with its limitations

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