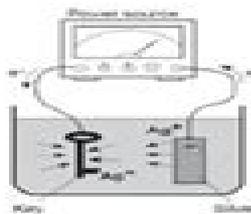


ENHANCING TEST ITEMS – CHEMISTRY (for Physical Sciences majors)

- Which of the following statements best describes the atomic radius of potassium atom and the atomic radius of a calcium atom?
 - The radius of the potassium atom is smaller because of its smaller nuclear charge.
 - The radius of the potassium atom is smaller because of its larger nuclear size.
 - The radius of the potassium atom is larger because its nuclear charge is smaller.
 - The radius of the potassium atom is larger because its nuclear charge is also larger.
 - Which type of chemical reaction occurs when an iron nail rusts?
 - condensation
 - neutralization
 - oxidation-reduction
 - ionization-dissociation
 - Which electron dot symbol represents the atom in period-4 with the highest first ionization energy?
 - 
 - 
 - 
 - 
 - Which phase change is exothermic?
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(g)}$
 - $\text{H}_2\text{O}_{(g)} \rightarrow \text{H}_2\text{O}_{(l)}$
 - $\text{H}_2\text{O}_{(l)} \rightarrow \text{H}_2\text{O}_{(s)}$
 - $\text{H}_2\text{O}_{(s)} \rightarrow \text{H}_2\text{O}_{(l)}$
 - Given the diagram, which statement BEST describes the key?
 - It is negative and acts as the cathode.
 - It is positive and acts as the cathode.
 - It is negative and acts as the anode.
 - It is positive and act as the anode
- 
- Which technique is safest for diluting a concentrated acid with water?
 - Add the acid to the water quickly.
 - Add the water to the acid quickly.
 - Add the acid to the water slowly while stirring constantly.
 - Add the water to the acid slowly while stirring constantly.
 - Which elements combine to form an ionic bond?
 - sodium and potassium
 - sodium and oxygen
 - carbon and oxygen
 - carbon and sulfur
 - As the elements in group 16 are considered in order from top to bottom, the atomic radius of each successive element increases. This increase is primarily due to an increase in
 - electronegativity
 - ionization energy
 - the number of protons occupying the nucleus
 - the number of occupied principal energy levels
 - An air pollution situation in which reactions occur on the surfaces of solid particles and produce acidic irritation of lung tissues is called _____.

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S Ashworth



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Rosenstein, 1916 **Chemistry Education and Sustainability in the Global Age** Mei-Hung Chiu, Hsiao-Lin Tuan, Hsin-Kai

Wu, Jing-Wen Lin, Chin-Cheng Chou, 2012-12-05 This edited volume of papers from the twenty first International Conference

on Chemical Education attests to our rapidly changing understanding of the chemistry itself as well as to the potentially enormous material changes in how it might be taught in the future Covering the full range of appropriate topics the book features work exploring themes as various as e learning and innovations in instruction and micro scale lab chemistry In sum

the 29 articles published in these pages focus the reader's attention on ways to raise the quality of chemistry teaching and learning promoting the public understanding of chemistry deploying innovative technology in pedagogy practice and research and the value of chemistry as a tool for highlighting sustainability issues in the global community Thus the

ambitious dual aim achieved in these pages is on the one hand to foster improvements in the teaching and communication of chemistry whether to students or the public and secondly to promote advances in our broader understanding of the subject that will have positive knock on effects on the world's citizens and environment In doing so the book addresses as did the

conference the neglect suffered in the chemistry classroom by issues connected to globalization even as it outlines ways to bring the subject alive in the classroom through the use of innovative technologies **Essays in Physical Chemistry**

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