

# Chemistry

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(Chapter 1)(Solutions) (InText Questions)  
(Class 12)

## Question 1.11:

Calculate the mass of ascorbic acid (Vitamin C,  $C_6H_8O_6$ ) to be dissolved in 75 g of acetic acid to lower its melting point by  $1.5^\circ\text{C}$ .  $K_f = 3.9 \text{ K kg mol}^{-1}$ .

## Answer 1.11:

Mass of acetic acid,  $w_1 = 75 \text{ g}$

Molar mass of ascorbic acid ( $C_6H_8O_6$ ),  $M_2 = 6 \times 12 + 8 \times 1 + 6 \times 16 = 176 \text{ g mol}^{-1}$

Lowering of melting point,  $\Delta T_f = 1.5 \text{ K}$

We know that:

$$\Delta T_f = \frac{K_f \times 1000 \times w_2}{M_2 \times w_1}$$

$$\Rightarrow w_2 = \frac{\Delta T_f \times M_2 \times w_1}{K_f \times 1000}$$

$$= \frac{0.5 \times 176 \times 75}{3.9 \times 1000}$$

$$= 5.08 \text{ g (approx)}$$

Hence, 5.08 g of ascorbic acid is needed to be dissolved.

## Question 1.12:

Calculate the osmotic pressure in Pascals exerted by a solution prepared by dissolving 1.0 g of polymer of molar mass 185,000 in 450 mL of water at  $37^\circ\text{C}$ .

## Answer 1.12:

It is given that:

Volume of water,  $V = 450 \text{ mL} = 0.45 \text{ L}$

Temperature,  $T = (37 + 273) \text{ K} = 310 \text{ K}$

Number of moles of the polymer,  $n = \frac{1}{185000} \text{ mol}$

We know that:  $\pi = \frac{n}{V} RT$

Osmotic pressure,

$$= \frac{1}{185000} \text{ mol} \times \frac{1}{0.45 \text{ L}} \times 8.314 \times 10^3 \text{ Pa L K}^{-1} \times 310 \text{ K}$$

$$= 30.98 \text{ Pa}$$

$$= 31 \text{ Pa (approximately)}$$

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# Chemistry For Today Answers And Solutions

**SJ Ball**



## **Chemistry For Today Answers And Solutions:**

## **Chemistry For Today Answers And Solutions** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has be evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Chemistry For Today Answers And Solutions**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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