

VSA QUESTIONS (1 - MARK QUESTIONS)

1. Define the term 'rate of reaction'.
2. Mention the units of rate of reaction.
3. Express the rate of reaction in terms of $\text{Br}^- (\text{aq})$ as reactant and $\text{Br}_2 (\text{aq})$ as product for the reaction :



4. For a chemical reaction represented by $\text{R} \rightarrow \text{P}$ the rate of reaction is denoted by

$$\frac{-\Delta[\text{R}]}{\Delta t} \text{ or } \frac{+\Delta[\text{P}]}{\Delta t}$$

Why a positive sign (+) is placed before $\frac{\Delta[\text{P}]}{\Delta t}$ and negative sign (-)

before $\frac{\Delta[\text{R}]}{\Delta t}$?

5. Express the rate of reaction in terms of disappearance of hydrogen and appearance of ammonia in the given reaction.



6. Why rate of reaction does not remain constant throughout?
7. Write the unit of first order rate constant of a gaseous reaction if the partial pressure of gaseous reactant is given in bar.
8. For a zero order reaction :

$\text{R} \rightarrow \text{P}$, the change in concentration of reactant w.r.t. time is shown by following graph.

Chemistry Chemical Kinetics Practice Questions Answers

Elizabeth Kean, Catherine Middlecamp

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Master the PCAT Peterson's, 2012-07-15 Peterson's Master the PCAT is an in depth review that offers thorough preparation for the computer based exam After learning about the structure format scoring and score reporting and the subtests and question types you can take a diagnostic test to learn about your strengths and weaknesses The next six parts of the eBook are focused on detailed subject reviews for each subtest verbal ability reading comprehension biology chemistry quantitative ability and writing Each review includes practice questions with detailed answer explanations You can take two practice tests to track your study progress The tests also offer detailed answer explanations to further improve your knowledge and understanding of the tested subjects The eBook concludes with an appendix that provides helpful information on a variety of careers in pharmacy and ten in depth career profiles [Kaplan SAT Subject Test Chemistry 2015-2016](#)

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