

## PRE TEST

## Chapter 6 Test-General/Honors Chemistry

## Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- |       |              |                           |                                                                                                              |
|-------|--------------|---------------------------|--------------------------------------------------------------------------------------------------------------|
| C.2.2 | <u>  a  </u> | 1. cation                 | a. a positively charged ion                                                                                  |
| C.2.2 | <u>  g  </u> | 2. electronegativity      | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled                     |
| C.2.5 | <u>  j  </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom                                                    |
| C.2.6 | <u>  i  </u> | 4. alkali earth metal     | d. good conductor of heat and electric current                                                               |
| C.2.6 | <u>  h  </u> | 5. noble gas              | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons     |
| C.2.5 | <u>  b  </u> | 6. representative element | f. any nonmetal in group 7A                                                                                  |
| C.2.2 | <u>  d  </u> | 7. metal                  | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>  f  </u> | 8. halogen                | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled            |
| C.2.5 | <u>  e  </u> | 9. transition metal       | i. an element in group 2A                                                                                    |
| C.2.2 | <u>  c  </u> | 10. ionization energy     | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons               |

## Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- C.2.6   b   11. The alkali metals do not include
- |       |       |
|-------|-------|
| a. Li | c. Na |
| b. Ca | d. Rb |
- C.2.6   c   12. The element iodine, I, is a
- |                          |                              |
|--------------------------|------------------------------|
| a. period 5 alkali metal | c. period 5 halogen          |
| b. period 4 halogen      | d. period 5 transition metal |

# Chemistry Chapter 6 Test

**M Walker**



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