

Atomic Target Practice

Solving the Structure of the Atom



Introduction

Rutherford scattering is one of the most famous experiments of all time. More than 25 years after conducting the experiment, Ernest Rutherford described the results this way:

"It was about as credible as if you had fired a 15-inch shell at a piece of tissue paper and it came back and hit you."

The experiment itself was actually the culmination of a series of experiments, carried out over a five-year period, dealing with the scattering of high-energy alpha particles by various substances. What is "Rutherford scattering" and why was it so important?

Concepts

- Atomic structure
- Atomic mass
- Nucleus
- Nuclear charge

Background

Ernest Rutherford received the Nobel Prize in Chemistry in 1908 for his investigations into the disintegration of the elements as a result of radioactive decay. Among the products of the radioactive decay of elements are alpha particles—small, positively charged, high-energy particles. In trying to learn more about the nature of alpha particles, Rutherford and his co-workers, Hans Geiger and Ernest Marsden, began studying what happened when a narrow beam of alpha particles was directed at a thin piece of metal foil. Alpha particles are a type of nuclear radiation, traveling at about 1/10 the speed of light. As expected for such high-energy particles, most of the particles penetrated the thin metal foil and were detected on the other side. What was unexpected was that a few—a very few, to be sure—of the alpha particles were actually reflected back toward the source, having been "scattered" or bent due to their encounters with the metal atoms in the foil target. The number of alpha particles that were reflected back depended on the atomic mass of the metal. Gold atoms, having the highest atomic mass of the metals studied, gave the largest amount of so-called "back scattering."

Rutherford's scattering experiments have been described as a "black box" experiment. The properties of the alpha particles, their mass, charge, speed, etc., were at least partially understood. The atoms making up the target, however, presented Rutherford with a kind of black box—the structure of the atom was not known at the time. In order to explain the results of the scattering experiments, Rutherford had to unlock the black box, that is, he had to solve the structure of the atom. In 1911 Rutherford proposed the following model for the structure of the atom:

- Most of the mass of the atom is concentrated in a very small, dense central area, later called the nucleus, which is about 1/100,000 the diameter of the atom.
- The rest of the atom is apparently "empty space." (*Most of the alpha particles traveled straight through the metal foil, as if nothing were in their path.*)
- The central, dense core of the atom is positively charged, with the nuclear charge equal to about one-half the atomic mass. (*A few alpha particles randomly struck the gold target, a few approached the nucleus of an atom head on. The positively charged alpha particles were strongly repelled by the nuclear charge and "recoiled" or bounced back to the source.*)

It is not practical to recreate Rutherford's original scattering experiments in the high school laboratory. Some idea of the challenge that faced Rutherford and his co-workers can be gained from the following "black box" activity using marbles and an unseen, unknown target.

Atomic Target Practice Lab Answers

M Lipman



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Understanding the properties and behavior of the COSMOS Don Hainesworth,2011 [In Search of Schrodinger's Cat](#) John Gribbin,2011-05-04 Quantum theory is so shocking that Einstein could not bring himself to accept it It is so important that it provides the fundamental underpinning of all modern sciences Without it we d have no nuclear power or nuclear weapons no TV no computers no science of molecular biology no understanding of DNA no genetic engineering In Search of Schrodinger s Cat tells the complete story of quantum mechanics a truth stranger than any fiction John Gribbin takes us step by step into an ever more bizarre and fascinating place requiring only that we approach it with an open mind He introduces the scientists who developed quantum theory He investigates the atom radiation time travel the birth of the universe superconductors and life itself And in a world full of its own delights mysteries and surprises he searches for Schrodinger s Cat a search for quantum reality as he brings every reader to a clear understanding of the most important area of scientific study today quantum physics In Search of Schrodinger s Cat is a fascinating and delightful introduction to the strange world of the quantum an essential element in understanding today s world **Nuclear Science Abstracts** ,1963 [Nuclear Science Abstracts](#) ,1974-07 **The EC Archives: Gunfighter Volume 1** Gardner Fox,2022-04-19 At last collecting the wild adventures of the lone lawman known as the Gunfighter and his comrade the Buckskin Kid as they take on multiple mean outlaws and ride a rough terrain of terror in this collection of thrilling pre trend EC tales of western lore This hardcover volume collects Gunfighter 5 9 recolored digitally using the original palette featuring art by legendary comics talents Johnny Craig Graham Ingels and more **Transactions of the American Nuclear Society** American Nuclear Society,1989 **The Annual American Catalog, 1900-1909** ,1906 *The Annual American Catalog, 1905* ,1906 **The Annual American Catalog** ,1906 *INIS Atomindex* ,1985 [The Publishers Weekly](#) ,1906 **U.S. Government Research & Development Reports** ,1967 **Science Illustrated** ,1948 **Government Reports Announcements & Index** ,1994 **Dissertation Abstracts International** ,1990 *South Western Reporter. Second Series* ,1992 [An Introductory Guide to EC Competition Law and Practice](#) Valentine Korah,1994 **Physics Briefs** ,1993 **Summaries of Papers Presented at the Conference on Lasers and Electro-optics** ,1988 *The Inclusive Classroom* Margo A. Mastropieri,Thomas E. Scruggs,2007 *The Inclusive Classroom Strategies for Effective Instruction Third Edition* By Margo A Mastropieri and Thomas E Scruggs Recent Praise for *The Inclusive Classroom Strategies for Effective Instruction* Quite detailed Quite practical Everything is well linked to both research and standards Joseph E Nolan Indiana University of Pennsylvania This text is exceptional The authors have provided numerous practical examples throughout the book that are easy for inexperienced teachers to implement I believe even teachers with extensive experience would find new ideas in this textbook Karen B Clark Indiana University South Bend As you can see the real strength of *The Inclusive Classroom* is its emphasis on practical research based teaching and learning strategies in an inclusive environment The third edition focuses

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