

Name: _____ Period: _____

Properties of Water Worksheet

Word Bank:

covalent solvent deposition dissolve cohesion surface cooler universal
tension polar viscosity condensation negatively adhesion positively sublimation

1. The hydrogen and oxygen atoms are held together by covalent bonds.
2. The electrons are not shared equally creating a polar molecule.
3. The polarity of water allows it to dissolve most substances. Because of this it is referred to as the universal solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positively charged hydrogen end of one water molecule attracts the negatively charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
7. surface tension creates the skin-like surface formed due to the polar nature of water.
8. viscosity is the tendency for fluids to resist flow.
cooler water is more viscous than warmer water, allowing floating organisms to use less energy to keep from sinking.
9. condensation is when water changes from a gas to a liquid.
10. sublimation is when water changes from a solid directly to a gas.
11. deposition is when water changes from a gas directly to a solid.
12. Why does ice float?
water is less dense in solid form because hydrogen bonding between water molecules pushed molecules further apart into a set crystalline pattern
13. List the properties that ocean water has due to the fact that it contains salt.
higher density
conductivity
decreased surface tension

Properties Of Water Lab Answers

Emilie Sanchez

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Table of Contents Properties Of Water Lab Answers

1. Understanding the eBook Properties Of Water Lab Answers
 - The Rise of Digital Reading Properties Of Water Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Properties Of Water Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Properties Of Water Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Properties Of Water Lab Answers
 - Personalized Recommendations

- Properties Of Water Lab Answers User Reviews and Ratings
- Properties Of Water Lab Answers and Bestseller Lists
- 5. Accessing Properties Of Water Lab Answers Free and Paid eBooks
 - Properties Of Water Lab Answers Public Domain eBooks
 - Properties Of Water Lab Answers eBook Subscription Services
 - Properties Of Water Lab Answers Budget-Friendly Options
- 6. Navigating Properties Of Water Lab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Properties Of Water Lab Answers Compatibility with Devices
 - Properties Of Water Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Properties Of Water Lab Answers
 - Highlighting and Note-Taking Properties Of Water Lab Answers
 - Interactive Elements Properties Of Water Lab Answers
- 8. Staying Engaged with Properties Of Water Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Properties Of Water Lab Answers
- 9. Balancing eBooks and Physical Books Properties Of Water Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Properties Of Water Lab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Properties Of Water Lab Answers
 - Setting Reading Goals Properties Of Water Lab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Properties Of Water Lab Answers
 - Fact-Checking eBook Content of Properties Of Water Lab Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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