

**UNIT 10: PERIODIC TABLE** **STANDARD CURRICULAR PRACTICE COMPETENCY: ANALYZING**

**Section 1.1 (continued)**

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the first two on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                            |
|---|----------------------------|
| _____ 18. A column in the periodic table                  | a. metals                  |
| _____ 19. A row in the periodic table                     | b. group                   |
| _____ 20. Group 1 elements                                | c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 17 elements                               | e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left of metals.
- \_\_\_\_\_ 26. Group 1A elements are the alkali metal metals.
- \_\_\_\_\_ 27. Group 1A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 1A elements are (not) nonmetallic elements located on nonmetallic metals.
- \_\_\_\_\_ 29. Alkali metals have properties of both metals and more nonmetallic metals.

# Chapter 7 Study Guide For Content Mastery Answer Key

**Edward G. Martin, Jerome S. Burstein**



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### **Table of Contents Chapter 7 Study Guide For Content Mastery Answer Key**

1. Understanding the eBook Chapter 7 Study Guide For Content Mastery Answer Key
  - The Rise of Digital Reading Chapter 7 Study Guide For Content Mastery Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Study Guide For Content Mastery Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 7 Study Guide For Content Mastery Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Study Guide For Content Mastery Answer Key

- Personalized Recommendations
  - Chapter 7 Study Guide For Content Mastery Answer Key User Reviews and Ratings
  - Chapter 7 Study Guide For Content Mastery Answer Key and Bestseller Lists
5. Accessing Chapter 7 Study Guide For Content Mastery Answer Key Free and Paid eBooks
    - Chapter 7 Study Guide For Content Mastery Answer Key Public Domain eBooks
    - Chapter 7 Study Guide For Content Mastery Answer Key eBook Subscription Services
    - Chapter 7 Study Guide For Content Mastery Answer Key Budget-Friendly Options
  6. Navigating Chapter 7 Study Guide For Content Mastery Answer Key eBook Formats
    - ePub, PDF, MOBI, and More
    - Chapter 7 Study Guide For Content Mastery Answer Key Compatibility with Devices
    - Chapter 7 Study Guide For Content Mastery Answer Key Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chapter 7 Study Guide For Content Mastery Answer Key
    - Highlighting and Note-Taking Chapter 7 Study Guide For Content Mastery Answer Key
    - Interactive Elements Chapter 7 Study Guide For Content Mastery Answer Key
  8. Staying Engaged with Chapter 7 Study Guide For Content Mastery Answer Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chapter 7 Study Guide For Content Mastery Answer Key
  9. Balancing eBooks and Physical Books Chapter 7 Study Guide For Content Mastery Answer Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chapter 7 Study Guide For Content Mastery Answer Key
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chapter 7 Study Guide For Content Mastery Answer Key
    - Setting Reading Goals Chapter 7 Study Guide For Content Mastery Answer Key
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chapter 7 Study Guide For Content Mastery Answer Key

- Fact-Checking eBook Content of Chapter 7 Study Guide For Content Mastery Answer Key
  - Distinguishing Credible Sources
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- Utilizing eBooks for Skill Development
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