

Acids and Bases Worksheet 1

1. Finish the following reactions when acids are added to water.



2. What are the conjugate bases of these acids?

original acid	conjugate base
HNO_3	
H_2O	
H_3O^+	
H_2SO_4	
HBr	
HCO_3^-	

3. What are the conjugate acids of these bases?

original base	conjugate acid
OH^-	
H_2O	
HCO_3^-	
SO_4^{2-}	
ClO_4^-	

4. Which of the following represent conjugate acid-base pairs?



5. Calculate the $[\text{H}^+]$ in a solution in which $[\text{OH}^-] = 2.0 \times 10^{-2} \text{ M}$. Is this solution acidic, neutral, or basic?

6. Calculate the $[\text{OH}^-]$ in a solution in which $[\text{H}^+] = 3.99 \times 10^{-5} \text{ M}$. Is this solution acidic, neutral, or basic?

Acids And Bases Study Guide Answer Key

Lingsheng Yao



Acids And Bases Study Guide Answer Key:

Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Acids And Bases Study Guide Answer Key**

In some sort of inundated with displays and the cacophony of immediate interaction, the profound power and emotional resonance of verbal artistry usually disappear into obscurity, eclipsed by the regular onslaught of sound and distractions. However, nestled within the musical pages of **Acids And Bases Study Guide Answer Key**, a captivating function of fictional elegance that pulses with organic emotions, lies an unique trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, that interesting opus courses viewers on a psychological odyssey, lightly revealing the latent possible and profound impact stuck within the complex internet of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is central styles, dissect its interesting writing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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