

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|---|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | the remaining substance when a proton is lost _____ |
| c. Conjugate Acid | _____ | from an acid _____ |
| d. Conjugate Base | _____ | a substance that produces OH ⁻ in solution _____ |
| e. Bronsted-Lowry Acid | _____ | a substance producing H ⁺ in solution _____ |
| f. Bronsted-Lowry Base | _____ | a proton acceptor _____ |
| | | a proton donor _____ |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- ; SO_4^{2-}
- HBr ; BrO^-
- H_3PO_4 ; PO_4^{3-}
- H_2O ; HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Acids Bases And Solutions Answer Key Lab35

Marco Casella



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Electrochemical Stripping Analysis František Vydra, Karel Štulík, Eva Juláková, 1976 Tsennaaks Stadii, bonjaaks slangii Heikki Paunonen, 1970 *Bulletin of Thermodynamics and Thermochemistry*, 1970 **Nachrichten aus Chemie, Technik und Laboratorium**, 1987 **Science Citation Index**, 1994 Vols for 1964 have guides and journal lists

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ACIDS AND BASES

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Acid-base Chemistry Clifford W. Hand, Harry Lyon Blewitt, 1986. Very Good. No Highlights or Markup. All pages are intact. *Acids, Bases and Salts* Brian J. Knapp, 1998. Acids, bases, and salts. chemlab. [Acid Bases in Analytical Chemistry](#) Izaak Maurits Kolthoff, 1964. **More Acids and Bases**, 1944. 1997, □□□□□□ □□□□

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