

NAME: \_\_\_\_\_

## Unit 1 Practice Exam 1

## 1. Atom Structure

Which of the following statements about atomic structure is **NOT** true?

- a) Most of the volume of an atom is empty space.
- b) Electrons are the fundamental particle with a negative charge.
- c) The number of protons and neutrons in an atom must be equal. **correct**
- d) Protons are about 2000 times larger than electrons.

## 2. Question Functional Relationships

Which of the following statements about functional relationship is **NOT** true?

- a) Plotting (1,2), (2,4), (3,6) is an inverse function. **correct**
- b) Plotting (1,1), (2,4), (3,9) is a parabolic function.
- c) Plotting (1,2), (2,4), (3,8), (4,16) is an exponential function.
- d) Plotting (1,1), (2,0.25), (3,0.11) is an inverse square function.

## 3. Scientific Notation Math

What is  $[(8 \times 10^{-5})(3 \times 10^9)]/(4 \times 10^{-5})$ ?

- a)  $6 \times 10^9$  **correct**
- b)  $6 \times 10^{-1}$
- c)  $9.6 \times 10^{10}$
- d)  $6.6 \times 10^9$

## 4. Approximations

The atomic weights of the halogens are listed below. If you rounded to the nearest integer to perform a calculation of molar mass, which would have the largest error?

- a) F: 19.0 g/mol
- b) Cl: 35.45 g/mol **correct**
- c) Br: 79.9 g/mol
- d) I: 126.9 g/mol

# Chemistry Midterm Lab Answers

**A Loxley**



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