



PYTHON

Dr. Abhishek Garg
Dr. Arshad Ali

The Python Blueprint

Python Unleashed: A Beginner's Guide to Coding Magic



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Blueprint Python Programming Manual

Mark Vasilkov



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web the volume of 1.00 mol of any gas at stp standard temperature 273.15 k and pressure 1 atm is measured to be 22.414 l we can substitute 101.325 kpa for pressure 22.414 l for volume and 273.15 k for temperature into the ideal gas equation and solve for r

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between
temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the
condition at which a phase change occurs from liquid to vapour or vapour to liquid

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
the volume and the number of particles in each flask indicates relative amounts