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## Student Exploration: pH Analysis

**Vocabulary:** acid, acidic, alkaline, base, indicator, neutral, pH

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

- Acids** are substances that produce hydrogen ions ( $H^+$ ) when dissolved in water. Lemon juice is an example of an acid.
  - What does lemon juice taste like? It tastes sharp and tangy.
  - What does it feel like if lemon juice gets in your eye? It would sting and burn your eyes.
- Bases** are substances that produce hydroxide ions ( $OH^-$ ) when dissolved in water. Hand soap is an example of a base.
  - What does soap feel like? It is slippery and smooth.
  - What does soap taste like? It tastes bitter.
  - What does it feel like if soap gets in your eye? It stings and burns your eyes.

### Gizmo Warm-up

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess  $H^+$  ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

The *pH Analysis* Gizmo allows you to find the pH of a variety of liquids. In the Gizmo, check that the **Substance in the tube** is **Ammonia**, and click **Test**. Wait until the animation is finished.



- Indicators** change color in acids or bases. What is the color of the pH paper? Dark Blue
- Compare the paper to the **pH color chart**. What is the pH of ammonia? 11
- Is ammonia acidic or alkaline? Alkaline

# Student Exploration Ph Analysis Answers

**K Morrison**



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## **Table of Contents Student Exploration Ph Analysis Answers**

1. Understanding the eBook Student Exploration Ph Analysis Answers
  - The Rise of Digital Reading Student Exploration Ph Analysis Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Ph Analysis Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Student Exploration Ph Analysis Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Ph Analysis Answers
  - Personalized Recommendations
  - Student Exploration Ph Analysis Answers User Reviews and Ratings
  - Student Exploration Ph Analysis Answers and Bestseller Lists
5. Accessing Student Exploration Ph Analysis Answers Free and Paid eBooks
  - Student Exploration Ph Analysis Answers Public Domain eBooks
  - Student Exploration Ph Analysis Answers eBook Subscription Services
  - Student Exploration Ph Analysis Answers Budget-Friendly Options
6. Navigating Student Exploration Ph Analysis Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Student Exploration Ph Analysis Answers Compatibility with Devices
  - Student Exploration Ph Analysis Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Student Exploration Ph Analysis Answers
  - Highlighting and Note-Taking Student Exploration Ph Analysis Answers
  - Interactive Elements Student Exploration Ph Analysis Answers

8. Staying Engaged with Student Exploration Ph Analysis Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Student Exploration Ph Analysis Answers
9. Balancing eBooks and Physical Books Student Exploration Ph Analysis Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Student Exploration Ph Analysis Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Ph Analysis Answers
  - Setting Reading Goals Student Exploration Ph Analysis Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Ph Analysis Answers
  - Fact-Checking eBook Content of Student Exploration Ph Analysis Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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