

MOLE TO MOLE STOICHIOMETRY WORKSHEET

Name : _____

Date : _____

1. Using the equation



a. Find the number of moles of Al used to produce 9.9 moles of H_2 .

b. Find the number of moles of HCl used to produce 1.11 moles of AlCl_3 .

c. Find the number of moles of AlCl_3 produced along with 0.0757 moles of H_2 .

2. The reaction between lithium and fluorine produces lithium fluoride. How many moles of lithium fluoride can be produced with 2.17 mol of fluorine?



3. In the reaction of copper (II) oxide breaking down to form copper and oxygen, how much copper is produced if 0.663 moles of oxygen is given off?



4. Fermentation is a complex chemical process of making wine by converting glucose into ethanol and carbon dioxide:



If 17.5 moles of ethanol were produced, how many moles of glucose were there initially?

5. Consider the reaction:



Calculate the moles of HCl needed to react completely with 8.25 moles of zinc.

Moles And Stoichiometry Practice Problems Answers

Matthew Johll



Moles And Stoichiometry Practice Problems Answers:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

A Visual Analogy Guide to Chemistry, 2e Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

Basics for Chemistry David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027

Improving Student Comprehension in Chemistry Laboratories Tracy Lynn Haroff, 2006

Chemistry Greg Curran, 2011 Covers all the topics in a typical one year high school chemistry curriculum

Chemistry John McMurry, 1998

Prentice Hall Chemistry, 2000

Holt Chemistry Ralph Thomas Myers, 2004

Modern Chemistry Nicholas D. Tzimopoulos, 1993

Chemistry Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives

General Chemistry James P. Birk, Donald M. Baird, 1994-03

Chemistry John Christian Bailar, 1984

R.R. Bowker's Software for Schools, 1987

Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04

Matthew Johll s Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same case driven approach that Johll uses in his acclaimed Investigating

Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book's specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more **General Chemistry** Brady,1990-01-02 *Chemistry* Paul B. Kelter,James D. Carr,Andrew Scott,1998 5 Steps to a 5 AP Chemistry, 2015 ed Richard H. Langley,2014-07-14 This easy to follow study guide includes a complete course review full length practice tests and access to online quizzes and an AP Planner app 5 Steps to a 5 AP Chemistry features an effective 5 step plan to guide your preparation program and help you build the skills knowledge and test taking confidence you need to succeed This fully revised edition covers the latest course syllabus and matches the new exam It also includes access to McGraw Hill Education's AP Planner app which will enable you to customize your own study schedule on your mobile device AP Planner app features daily practice assignment notifications delivered to your mobile device 2 full length practice AP Chemistry exams Access to online AP Chemistry quizzes 3 separate study plans to fit your learning style **5 Steps to a 5 AP Chemistry, 2014-2015 Edition** Richard H. Langley,John Moore,2013-07-10 A PERFECT PLAN for the PERFECT SCORE STEP 1 Set up your study plan with three customized study schedules STEP 2 Determine your readiness with an AP style diagnostic exam STEP 3 Develop the strategies that will give you the edge on test day STEP 4 Review the terms and concepts you need to score high STEP 5 Build your confidence with full length practice exams *A Problem-Solving Approach to Aquatic Chemistry* James N. Jensen,2003-03-13 This text provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics Software designed especially for the text allows the reader to build complex models by applying equilibrium calculation principles Important features include material specific and integrated case studies thought provoking questions key ideas and historical sketches

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