

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Modern Chemistry Chapter 8 Section 2 Answers

Tom Palko



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Lab Experiments Modern Chemistry Harold Clark Metcalfe,1982 **Modern Chemistry** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 *Technical World Magazine* ,1913 The Robert Morris Associates Monthly Bulletin ,1927 The Annual American Catalogue Cumulated ,1903 **Lumber Manufacturer and Dealer** ,1914 *What Does this Mean?* James W. Voelz,1997 What does this mean is a seminary level introduction to hermeneutics drawing especially on the Lutheran theological tradition to define and shape the tasks of interpretation In parts 1 and 2 Voelz describes the major issues and methods of textual criticism semantics and pragmatics In the final section he offers insights for reading Scripture from a confessional position **The Chautauquan** ,1889 **The Lumber Manufacturer and Dealer** ,1915

Chemical Engineering Catalog ,1919 *Te HS&T J* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02 **Modern Techniques in Computational Chemistry: MOTECC-91** E. Clementi,1991-07-31

Instructor's Manual to Accompany Chemistry in the Modern World, Concepts and Applications Frank L. Wiseman,1985 Laboratory Procedures for the Medical Office Tom Palko,1996 **Church & school gazette** Church and school gazette,1868 Biochemistry John T. Tansey,2019-04-02 Biochemistry An Integrative Approach is addressed to premed biochemistry and life science majors taking a one semester biochemistry course This version includes the first 12 chapters and should only be used for one semester biochemistry courses Biochemistry addresses the diverse needs of premed biochemistry and life science majors by presenting relevant material while still preserving a chemical perspective Presented within the next generation of WileyPLUS Biochemistry emphasizes worked problems through video walkthroughs interactive elements and expanded end of chapter problems with a wide range of subject matter and difficulty The worked problems in the course are both qualitative and quantitative and model for students the biochemical reasoning they need to practice Students will often be asked to analyze data and make critical assessments of experiments The Pharmaceutical Journal and Pharmacist ,1909 **Pharmaceutical Journal** ,1909 *The World of Biology* P. William Davis,Eldra Pearl Solomon,Linda R. Berg,1990 Includes bibliographical references and index *Decorator* ,1925

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