
Biology 101 Exam 1 With 100% Correct Answers 2023

covalent bond - Correct Answer-The sharing of a pair of valence electrons by two atoms. (Ex: Hydrogen, oxygen, water, Methane)

Polar covalent bond - Correct Answer-When electrons are not equally shared among the atoms within a bond. (Ex: water) This causes a negative and a positive side of a molecule.

Hydrogen Bond - Correct Answer-A type of weak chemical bond formed when the slightly positive hydrogen atom of a polar covalent bond in one molecule is attracted to the slightly negative atom of a polar covalent bond in another molecule.

Acid - Correct Answer-a solution with high amounts of H⁺ (pH 0-6)

Base - Correct Answer-a solution with low amounts of H⁺ (pH 8-14)

Neutral solution - Correct Answer-a solution with equal parts H⁺ and OH⁻; pure water (pH ~ 7)

Buffer - Correct Answer-a substance that minimizes changes in pH. (Carbonic acid)

Hydroxyl group - Correct Answer-(-OH) can form hydrogen bonds with water , helps dissolve sugars

POLAR POLAR

Carboxyl group - Correct Answer-(-COOH) organic acid, found in acids i.e. vinegar. is

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Amino Group - Correct Answer-basic, can pick up and H⁺, In an amino acid

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Sulfhydryl group (-SH) - Correct Answer-(-SH) Resemble hydroxyl, POLAR POLAR

Phosphate group - Correct Answer-Backbone for phospholipids, most prevalent molecules in cell membranes

ATP - Correct Answer-adenosine triphosphate - 3 phosphate groups plus an adenosine, used for energy within cells. contains a ribose sugar.

Biology 101 Test And Answers

William D. Hedges



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Table of Contents Biology 101 Test And Answers

1. Understanding the eBook Biology 101 Test And Answers
 - The Rise of Digital Reading Biology 101 Test And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Biology 101 Test And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Biology 101 Test And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Biology 101 Test And Answers
 - Personalized Recommendations
 - Biology 101 Test And Answers User Reviews and Ratings
 - Biology 101 Test And Answers and Bestseller Lists
5. Accessing Biology 101 Test And Answers Free and Paid eBooks

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 - Biology 101 Test And Answers Budget-Friendly Options
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 - Highlighting and Note-Taking Biology 101 Test And Answers
 - Interactive Elements Biology 101 Test And Answers
8. Staying Engaged with Biology 101 Test And Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Biology 101 Test And Answers
9. Balancing eBooks and Physical Books Biology 101 Test And Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Biology 101 Test And Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Biology 101 Test And Answers
- Setting Reading Goals Biology 101 Test And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Biology 101 Test And Answers
- Fact-Checking eBook Content of Biology 101 Test And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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