

REPORT SHEET
Acid-Base Titration

LAB
10

A. Concentration of Acetic Acid in Vinegar

1. Brand Heinz Volume 5.00 mL (% on label) 5.0 %
 2. Molarity (M) of NaOH _____ M

	Trial 1	Trial 2	Trial 3
3. Initial NaOH level in buret	0.1 mL	0.2 mL	0.1 mL
4. Final NaOH level in buret	39.6 mL	41.2 mL	40.1 mL
5. Volume (mL) of NaOH used			
6. Average volume (mL)			
7. Average volume in liters (L)			

Show all calculations for the following questions.

8. Moles of NaOH used in titration _____ mole NaOH
9. Moles of $\text{HC}_2\text{H}_3\text{O}_2$ neutralized by NaOH _____ mole $\text{HC}_2\text{H}_3\text{O}_2$
10. Molarity of $\text{HC}_2\text{H}_3\text{O}_2$ _____ M $\text{HC}_2\text{H}_3\text{O}_2$
11. Grams of $\text{HC}_2\text{H}_3\text{O}_2$ _____ g $\text{HC}_2\text{H}_3\text{O}_2$
12. Percent (m/v) $\text{HC}_2\text{H}_3\text{O}_2$ in vinegar _____ % $\text{HC}_2\text{H}_3\text{O}_2$

Acid Base Titration Curve Lab Answers

**Seung-Beom Hong, M. Bazlur
Rashid, Lory Z. Santiago-Vázquez**

Acid Base Titration Curve Lab Answers:

Computer Based Projects for a Chemistry Curriculum Thomas J. Manning, Aurora P. Gramatges, 2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level. The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling. Each project is explained in a simple easy to understand manner. The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises. Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom.

A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2022-12-20 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry. The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics. The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases. The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points. To aid in reader comprehension important terms are defined in context and key ideas are summarized. Many thought provoking discussion questions worked examples and end of chapter problems are also included. Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter. In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text. A Problem Solving Approach to Aquatic Chemistry includes Fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria. Solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques. Acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity. Complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry. Oxidation reduction equilibria including equilibrium calculations graphical approaches and applications. Gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change. Other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies. For advanced senior undergraduates and first year graduate students in environmental engineering courses A Problem Solving Approach to Aquatic Chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings.

Methods in Biotechnology Seung-Beom Hong, M. Bazlur Rashid, Lory Z. Santiago-Vázquez, 2016-05-12 As rapid

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