

# STOICHIOMETRY LIMITING REAGENT

- ① Consider the following reaction:



Balance the reaction.

- a 200 grams of each reactant react. How many grams of silver phosphate are formed?

- b How much excess reagent is left?

- ② Consider the following reaction:



Balance the reaction.

How many grams of  $\text{NH}_3$  are produced when 28 g of  $\text{N}_2$  reacts with 25 g of  $\text{H}_2$ ? What is the limiting reagent? What is the excess reagent? How many grams of excess reagent remain?

# Stoichiometry Limiting Reagent Problems Answers

**S Marginson**



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