

Series Circuits

READ

In a series circuit, current follows only one path from the positive end of the battery toward the negative end. The total resistance of a series circuit is equal to the sum of the individual resistances. The amount of energy used by a series circuit must equal the energy supplied by the battery. In this way, electrical circuits follow the law of conservation of energy. Understanding these facts will help you solve problems that deal with series circuits.

To answer the questions in the practice section, you will have to use Ohm's law. Remember that:

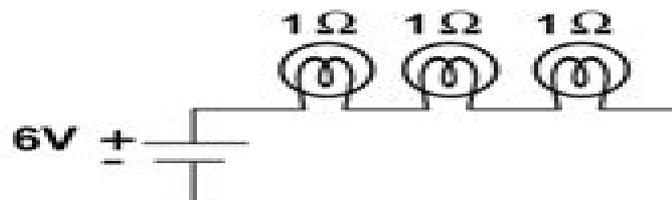
$$\text{Current (amps)} = \frac{\text{Voltage (volts)}}{\text{Resistance (ohms)}}$$

Some questions ask you to calculate a *voltage drop*. We often say that each resistor (or light bulb) creates a separate voltage drop. As current flows along a series circuit, each resistor uses up some energy. As a result, the voltage gets lower after each resistor. If you know the current in the circuit and the resistance of a particular resistor, you can calculate the voltage drop using Ohm's law.

$$\text{Voltage drop across resistor (volts)} = \text{Current through resistor (amps)} \times \text{Resistance of one resistor (ohms)}$$

PRACTICE

- Use the series circuit pictured to the right to answer questions (a)-(e).
 - What is the total voltage across the bulbs?
 - What is the total resistance of the circuit?
 - What is the current in the circuit?
 - What is the voltage drop across each light bulb? (Remember that voltage drop is calculated by multiplying current in the circuit by the resistance of a particular resistor: $V = IR$.)
 - Draw the path of the current on the diagram.
- Use the series circuit pictured to the right to answer questions (a)-(e).
 - What is the total voltage across the bulbs?
 - What is the total resistance of the circuit?
 - What is the current in the circuit?
 - What is the voltage drop across each light bulb?
 - Draw the path of the current on the diagram.
- What happens to the current in a series circuit as more light bulbs are added? Why?
- What happens to the brightness of each bulb in a series circuit as additional bulbs are added? Why?



Circuits Series And Parallel Answer Key

RD Boyd



Circuits Series And Parallel Answer Key:

Electricity and Magnetism Edward P. Ortleb, Richard Cadice, 1991-09-01 Color Overheads Included This book presents a program of basic studies dealing with electricity and magnetism Properties and types of electricity and different methods of producing electricity are detailed Information is provided on motors and other appliances that use electricity Each of the twelve teaching units in this book is introduced by a color transparency which emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key

Electronics Technology Fundamentals Robert T. Paynter, B. J. Toby Boydell, 2005
Electronics Technology Fundamentals is a complete introduction to the increasingly complex study of electronics This text presents do circuits ac circuits and devices in one condensed easy to read volume allowing these fundamentals to be covered in less time than required by traditional texts Hailed by instructors as an excellent innovative approach to teaching the fundamentals the text presents all of the same vital information offered in traditional books while implementing the engaging clear writing style and superb learning tools developed by seasoned authors Robert T Paynter and B J Toby Boydell The following features are NEW to this Second Edition Full 4 color format improving clarity and visual appeal Chapter opening vignettes helping the reader to connect the chapter material to real world circuits and applications New sections introducing the reader to component testing and fault symptoms Many newer components and component packages appearing throughout New margin notes introducing applications of principles and circuits New margin notes demonstrating calculator key sequences for many of the problem solving examples

[Electricity & Magnetism \(eBook\)](#) Edward P. Ortleb, Richard Cadice, 1991-09-01 This book presents a program of basic studies dealing with electricity and magnetism Properties and types of electricity and different methods of producing electricity are detailed Information is provided on motors and other appliances that use electricity Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key

[1500 Science Test Questions/Answers](#) Dennis Arden Hooker, 2025-01-01 1500 Science Test Questions w Keys Answers Statistical Analysis For Science Teachers Upper Elementary to College Dr Hooker researched and developed a book of 1500 Science Test Questions together with the Bloom s Taxonomy Discrimination Index the Key etc The book was funded through the National Science Foundation for teachers of Upper Middle School through College Science Programs 1500 Science Test Questions is an excellent tool for teachers to develop their own tests and for students to study for High School and College proficiency exams

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The Railway Signal Dictionary Braman B.. Adams, Rodney Hitt, 1912 **Official Gazette of the United States Patent Office** USA Patent Office, 1927 **Circuit Analysis** Allan Robbins, Wilhelm C. Miller, 1995 This ABET level optional calculus introduced emphasis on problem solving introductory DC AC text covers electrical circuit theory beginning with foundational theorems and basic DC concepts and advancing through to AC topics

Electricity and Magnetism, 2013

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Sound Waves, 1905 **Mastering Electricity** Stuart

Asser, Vincent Stigliano, Stuart M. Asser, 1993 Each chapter includes a list of key terms and used in the material and a set of student objectives that a student should be able to meet after completing the section TechTidbits T2 and Tech Troubleshooting T3 clarifying notes and safety points are included throughout ALSO AVAILABLE Laboratory Manual ISBN 0 8273 4603 4 INSTRUCTOR SUPPLEMENTS CALL CUSTOMER SUPPORT TO ORDER Computerized Testbank ISBN 0 8273 5816 4 Printed Testbank ISBN 0 8273 5817 2 Instructor s Resource Manual ISBN 0 8273 4605 Keywords Survey Electronics **Florida Schools** ,1964 **Series-parallel Circuits** ,1984 **Principles of Electricity, for Students of Physics and Engineering** Eugene George Key,1967

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and

percent yield of this reaction will be calculated

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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stirring the solution adding water then removing water adding acetone heating the acetone on low

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