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## Reading Counts for Students Overview

# Reading Counts Test Answers

**S Ashworth**



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web thermal energy from the hotter object to the cooler object the hotter object will cool down and the cooler one will warm up until they are at the same temperature where there will be no net energy transfer two objects are in thermal equilibrium when there is no net transfer of thermal energy between them solids liquids and gases

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many different places such as the sun the earth's core and even in people thermal energy is used for many different purposes such as heating homes and generating electricity

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web a common misconception is that heat and thermal energy are the same thing heat is like work it is a means for transferring energy not the energy itself though it has units of energy that is heat is not contained within a system like thermal energy is

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ft i c in the equation  $q$  stands for the change in thermal energy

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web thermal energy is defined as when any atoms or molecules of an element start vibrating due to the rise in temperature it produces energy which is called thermal energy in other words thermal energy definition can be the movement of atoms and molecules these atoms and molecules will keep moving faster as the temperature rises

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