

# Types of Cooking Methods



**Boiling**



**Simmering**



**Steaming**



**Baking**



**Grilling**



**Broiling**



**Shallow frying**



**Pan-frying**



**Stewing**



**Poaching**



**Blanching**



**Roasting**



**Sautéing**



**Stir-frying**



**Deep-frying**



**Smoking**



**Sous vide**



**Pressure cooking**

# Cooking Techniques Manual Complete Workbook

**J Ma**



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was produced i what mass of iodine was produced

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g Al x 1 mol Al x 2 mol  $\text{Al}_2\text{O}_3$  x 101.964 g  $\text{Al}_2\text{O}_3$  302.3 g  $\text{Al}_2\text{O}_3$  26.982 g Al 4 mol Al 1 mol  $\text{Al}_2\text{O}_3$  c determine the percent yield  
yield actual x 100 260.0 g x 100 86.01 theoretical 302.3 g

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 $\text{CH}_3\text{OH}$  can be produced by the following reaction  $2\text{H}_2 + \text{CO} \rightarrow \text{CH}_3\text{OH}$  calculate the theoretical yield of  $\text{CH}_3\text{OH}$  if 68.5 g of CO is  
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