

STOICHIOMETRY PROBLEMS

1. The combustion of a sample of butane, C_4H_{10} (lighter fluid), produced 2.46 grams of water.



- (a) How many moles of water formed?
- (b) How many moles of butane burned?
- (c) How many grams of butane burned?
- (d) How much oxygen was used up in moles?
- (e) How much oxygen was used up in grams?

2. Sodium metal burns in air according to the balanced reaction below.



- (a) How many moles of oxygen are needed to react with 9.5 g of sodium completely?
- (b) How many grams of sodium are needed to produce 12.5 g of sodium oxide?

3. In the combustion of 54.5 g of butane (C_4H_{10}), how many grams of CO_2 are produced? Write and balance the equation before solving.

Stoichiometry 8b Extra Practice Problems Answers

Christian Drosten



Stoichiometry 8b Extra Practice Problems Answers:

Enjoying the Beat of Term: An Mental Symphony within **Stoichiometry 8b Extra Practice Problems Answers**

In a world eaten by screens and the ceaseless chatter of fast interaction, the melodic splendor and emotional symphony created by the written term often fade in to the backdrop, eclipsed by the constant sound and distractions that permeate our lives. Nevertheless, situated within the pages of **Stoichiometry 8b Extra Practice Problems Answers** an enchanting fictional treasure full of raw feelings, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, that charming masterpiece conducts visitors on a mental trip, well unraveling the concealed tunes and profound impact resonating within each carefully constructed phrase. Within the depths of this touching review, we will explore the book is central harmonies, analyze its enthralling writing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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