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## Miller & Levine Biology Chapter 2 Questions and Answers

atoms - ANS-the building blocks of matter

proton - ANS-positively charged part of an atom in the nucleus

neutron - ANS-part of atom with no charge in the nucleus

electron - ANS-negatively charged particle of an atom that orbits around the nucleus and is 1/1840 the size of a proton

element - ANS-a pure substance with only one type of atom

compound - ANS-two or more elements chemically combined

ionic bonds - ANS-when one or more electrons are transferred from one atom to the other

covalent bonds - ANS-when electrons are shared between two atoms

molecule - ANS-smallest unit of most compounds

polar - ANS-one end is positive, the other is negative

hydrogen bonds - ANS-hold water molecules, not as strong as chemical bonds but strongest a molecule can form

cohesion - ANS-attraction between molecules of the same substance

adhesion - ANS-attraction between molecules of different substances

mixture - ANS-2 or more elements or compounds physically mixed but not chemically combined and it can be separated

solution - ANS-water and a dissolved substance

solute - ANS-what is being dissolved

solvent - ANS-what it is being dissolved in

suspension - ANS-water and a non-dissolved material

pH scale - ANS-scale that describes how acidic or basic a substance is

# Biology Chapter 14 Assessment Answers

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Test Bank for Starr's Biology, 2002

**College Board Achievement Test in Biology** Lawrence Solomon, 1983-12 Sample tests provide a review of aspects of biology such as cell structure reproduction genetics evolution biochemistry and the nervous system

Kaplan GRE Subject Test: Biology Kaplan, Kaplan Publishing, 2010-06-29 This comprehensive guide features targeted review of the concepts tested on the exam from cellular structure and molecular biology to ecology and evolution

*An Introduction to Methods & Models in Ecology, Evolution, & Conservation Biology* Stan Braude, 2010 Populations demography genetics

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