

# CHIMICA GENERALE

Ogni atomo/elemento è caratterizzato da una simbolo.

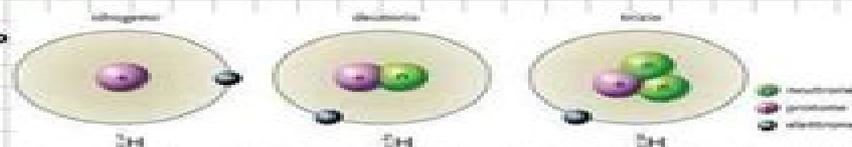


- il numero di massa è dato dalla somma di protoni + neutroni (nucleoni)
- il numero atomico corrisponde al numero dei protoni (uguale al numero di e<sup>-</sup>)

un nucleo con particolari valori di Z e A → **NUCLIDE**

**GLI ISOTOP:**

Atomi di uno stesso elemento che possono differire per il numero di neutroni. I diversi isotopi di un elemento hanno uguali proprietà chimiche, ma diverse proprietà fisiche.



→ **CATIONI**: quando ad un atomo vengono sottratti uno o più elettroni, si formano ioni carichi positivamente.

→ **ANIONI**: quando l'atomo assume uno o più elettroni si formano ioni carichi negativamente.

- contiene 7 periodi (orizzontali)
- 18 gruppi (verticali) → nei gruppi gli elementi hanno caratteristiche chimiche simili

**TAVOLA DEGLI ELEMENTI:**

essi sono organizzati per massa atomica crescente.

gruppi	1A	2A	3A	4A	5A	6A	7A	8A	9A	10A	11A	12A	13A	14A	15A	16A	17A	18A
1	H	He																
2	Li	Be																
3	Na	Mg	Al	Si	P	S	Cl	Ar										
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra																

■ orbitale s  
■ orbitale p  
■ orbitale d  
■ orbitale f

- **GRUPPO 1**: metalli alcalini (Li, Na, K, Rb, Cs)
- **GRUPPO 2**: metalli alcalino-terrosi (Be, Mg, Ca, Sr, Ba)
- essi reagiscono con l'acqua producendo idrogeno e calore es. il potassio reagisce tanto da fiammare l'idrogeno.
- **GRUPPO 18**: gas nobili, essi si combinano con pochissimi altri elementi. Sono tutti gas incolori e inodori. Sono poco reattivi.
- **GRUPPO 17**: alogeni. Si presentano con molecole biatomiche.

- **I METALLI**: sono solidi, dotati di lucentezza, malleabilità e duttilità; essi sono buoni conduttori di elettricità e calore.

- **I NON METALLI**: sono gas o solidi che non hanno caratteristiche metalliche.

- **SEMIMETALLI O METALLOIDI**: hanno caratteristiche intermedie tra i metalli e i non metalli.

# Appunti Di Chimica Generale Unife

**Joacim Rocklöv**



## **Appunti Di Chimica Generale Unife:**

## **Appunti Di Chimica Generale Unife** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Appunti Di Chimica Generale Unife**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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