

# AP CHEMISTRY FORMULA SHEET 2025

| Equations  | Constants   | Standard Reduction Potentials   |  |
|--|---|---|--|
| $PV = nRT$<br>$\ln[A] = -kt + \ln[A]_0$<br>$1/A = kt + 1/[A]_0$<br>$A = abc$<br>$\Delta G = -\Delta H - T\Delta S$<br>$q = mc\Delta T$<br>$q = -q$<br>$E = hy$<br>$pH = -\log[H^+]$<br>$pOH = -\log[OH^-]$<br>$pH + pOH = 14$<br>$K_a = [H^+][A^-]/[HA]$<br>$K_w = [H^+][OH^-] = 1.0 \times 10^{-14}$<br>$pK_a = -RT \ln K$<br>$pH = pK_a + \log [A^-]/[HA]$ | $R = 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K}$<br>$0.314 \text{ J/mol}\cdot\text{K}$<br>$N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$<br>$F = 96,485 \text{ C/mol}\cdot\text{e}^-$<br>$c = 3.00 \times 10^8 \text{ m/s}$<br>$h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}$<br>$P = 1 \text{ atm}$<br>$T = 298 \text{ K (25}^\circ\text{C)}$ | $\text{Cu}^{2+} + 2\text{e}^- \rightarrow \text{Cu}$<br>$\text{Zn}^{2+} + 2\text{e}^- \rightarrow \text{Zn}$<br>$\text{Cl}_2 + 2\text{e}^- \rightarrow 2\text{Cl}^-$<br>$2\text{H}_2\text{O} + 2\text{e}^- \rightarrow \text{H}_2 + 2\text{OH}^-$<br>$\text{Fe}^{3+} + \text{e}^- \rightarrow \text{Fe}^{2+}$ | $\text{Cu}^{2+} \rightarrow \text{Cu} \quad E$<br>$\text{Zn}^{2+} \rightarrow \text{Zn} \quad 0.34$<br>$\text{Cl}_2 \rightarrow 2\text{Cl}^- \quad 1.36$<br>$\rightarrow \text{K} \quad 0.83$<br>$\text{Mg}^{2+} \rightarrow \text{Mg} \quad 2.37$ |
|  |   | Thermochemistry Data  |  |
|  | <b>Thermochemistry Data</b><br>$\Delta H_f^\circ$ : kJ/mol<br>Bond energies: kJ/mol<br>Specific heat capacity, $C_p$ : J/g $^\circ\text{C}$   | $\Delta H^\circ$ : kJ/mol<br>Bond energies: kJ/mol<br>Specific heat capacity, $C_p$ : J/g $^\circ\text{C}$  | $\text{NO}_3^-$<br>$\text{SO}_4^{2-}$<br>$\text{PO}_4^{3-}$<br>Ammonium<br>$\text{OH}^-$<br>$\text{CO}_3^{2-}$ carbonate<br>$\text{CH}_3\text{COO}^-$ or $\text{C}_2\text{O}_4^{2-}$   |

# Ap Chemistry Chapter 7

**JS Bruner**



## Ap Chemistry Chapter 7:

## Enjoying the Tune of Appearance: An Mental Symphony within **Ap Chemistry Chapter 7**

In a global used by displays and the ceaseless chatter of immediate connection, the melodic elegance and psychological symphony developed by the prepared term usually disappear into the backdrop, eclipsed by the constant noise and distractions that permeate our lives. But, located within the pages of **Ap Chemistry Chapter 7** an enchanting literary treasure brimming with organic emotions, lies an immersive symphony waiting to be embraced. Crafted by an elegant musician of language, that fascinating masterpiece conducts readers on a mental journey, skillfully unraveling the hidden melodies and profound impact resonating within each cautiously crafted phrase. Within the depths of this moving review, we can examine the book is key harmonies, analyze their enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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