

**Atomic Orbitals**

Use with Chapter 5,  
Section 5.2

1. What is the shape of an s orbital?  
Spherical
2. What is the relationship between the size of an s orbital and the principal energy level in which it is found?  
\_\_\_\_\_
3. What is the shape of a p orbital? How many p orbitals are there in a sublevel?  
\_\_\_\_\_
4. How many electrons can each orbital hold?  
2
5. Look at the diagrams of the p orbitals. What do x, y, and z refer to?  
\_\_\_\_\_
6. How many d orbitals are there in a given sublevel? How many total electrons can the d orbitals in a sublevel hold?  
5 10
7. Which d orbitals have the same shape?  
 $x^2 - y^2$
8. What point in each diagram represents an atom's nucleus?  
\_\_\_\_\_
9. How likely is it that an electron occupying a p or a d orbital would be found very near an atom's nucleus? What part of the diagram supports your conclusion?  
\_\_\_\_\_

# Teaching Transparency 16 Answers

**Minjie Lin**



**Teaching Transparency 16 Answers:**

## **Teaching Transparency 16 Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Teaching Transparency 16 Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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