

Fuel Cells Harvest Chemical Energy

6

The real world applications become even more interesting when oxygen for energy delivery can come from a liquid electrolyte. Fuel cells can be made with a liquid proton conductor. Membranes for this purpose, such as proton exchange membranes, are made from acid ion-exchange resins. These resins are made with sulfonic acid (SA) groups. The SA groups in these resins are able to donate protons to the membrane. In this way, the SA groups in the membrane are able to donate protons, and in doing so, the membrane can be used to generate electricity. This type of application has been used to power the space shuttle and other applications.

Applying Your Knowledge

Answer the questions below and complete additional responses to the questions below.

1. What are the two main components of a fuel cell? How do they work together to generate electricity? (2 points)

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R Bogdan



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