

CHAPTER 24 NUCLEAR

# Nuclear Chemistry

## SECTION 1

Write a balanced nuclear equation for each reaction in the space provided.

1. \_\_\_\_\_ How many alpha particles are emitted in the decay of  $^{238}\text{U}$  to  $^{14}\text{C}$ ?  
 a. 6  
 b. 8  
 c. 10  
 d. 12
2. \_\_\_\_\_ The  $\beta$ -decay of  $^{14}\text{C}$  produces  $^{14}\text{N}$ .  
 a.  $^{14}\text{C}$  is oxidized to  $^{14}\text{N}$ .  
 b.  $^{14}\text{C}$  is reduced to  $^{14}\text{N}$ .  
 c.  $^{14}\text{C}$  is oxidized to  $^{14}\text{N}$ .  
 d.  $^{14}\text{C}$  is reduced to  $^{14}\text{N}$ .
3. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
4. \_\_\_\_\_  $^{238}\text{U}$  decays to  $^{206}\text{Pb}$  by emitting \_\_\_\_\_.  
 a.  $\alpha$  particles  
 b.  $\beta$  particles  
 c.  $\gamma$  rays  
 d.  $\alpha$  particles and  $\beta$  particles
5. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
6. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
7. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
8. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
9. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
10. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
11. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
12. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
 b. 147  
 c. 148  
 d. 149
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 a. 146  
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 a. 146  
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20. \_\_\_\_\_ The number of neutrons in  $^{238}\text{U}$  is \_\_\_\_\_.  
 a. 146  
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 d. 149

# Chapter 22 Review Nuclear Chemistry Answer Key

**L Reisser**



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