

Name :

Date :

Periodic Table Worksheet

1. What do you call the element series from atomic number 57 to 71?

2. Which among the following elements has a symbol that starts with a letter different from the first one in its name: aluminum, copper, gold, and rhenium?

3. What is the first element with an atomic mass greater than 100?

4. What is the first basic metal on the periodic table?

5. Where are the most active metals located?

6. Where are the most active nonmetals located?

7. Where is the highest electronegativity found?

8. Where is the lowest electronegativity found?

9. What is the atomic symbol for silver?

10. What is the last element in period 4?

Teaching Transparency Worksheet The Periodic Table Answers

Edward Shevick



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Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Chemistry in the Community** American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

Atoms, Molecules & Elements: Properties of Important Elements Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice Properties of Important Elements from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

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Mastering the Periodic Table Linda Trombley, J. Weston Walch (Firm), 1985 *Atoms,*

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Elements and the Periodic Table, Grades 5 - 12 Theodore S. Abby, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science

Facilitating Conceptual Change in Students' Understanding of the Periodic Table Mansoor Niaz, Marniev Luiggi, 2013-07-25 This book is about how students are taught the periodic table It reviews aspects of the periodic table s development using the history and philosophy of science The teaching method presented in this book is ideal for teaching the subject in high school and at introductory university level Chemistry students taught in this new experimental way are compared with those taught in the traditional way and the author describes how tests found more conceptual responses from the experimental group than the control group The historical aspects of importance to this teaching method are the role of the Karlsruhe Congress of 1860 the accommodation of the chemical elements in the periodic table prediction of elements that were discovered later corrections of atomic weights periodicity in the periodic table as a function of the atomic theory and the accommodation of argon The experimental group of students participated in various activities including discussion of various aspects related to the history and philosophy of science construction of concept maps and their evaluation by the students PowerPoint presentations and interviews with volunteer students

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