

**TEACHING TRANSPARENCY WORKSHEET**

**9**

# Types of Matter

**Use with Chapter 3,  
Section 3.4**

**1.** Into what two broad classes can all matter be divided?  
\_\_\_\_\_

**2.** What is the difference between a mixture and a pure substance?  
\_\_\_\_\_

**3.** What is the difference between a compound and an element?  
\_\_\_\_\_

**4.** Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.  
\_\_\_\_\_

**5.** A list of compounds and elements is given below. Circle the substances that are elements.

- |       |          |          |
|-------|----------|----------|
| gold  | aluminum | water    |
| sugar | oxygen   | platinum |
| salt  | chlorine | brass    |

**6.** How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?  
\_\_\_\_\_

**7.** Label each mixture below as either homogeneous or heterogeneous.

- |   |                              |
|---|------------------------------|
| <b>a.</b> air _____                           | <b>e.</b> finger paint _____ |
| <b>b.</b> clay _____                          | <b>f.</b> vinegar _____      |
| <b>c.</b> homemade lemonade (with pulp) _____ | <b>g.</b> soil _____         |
| <b>d.</b> oatmeal raisin cookie _____         |                              |

**8.** List three methods that are commonly used to separate mixtures into their component substances.  
\_\_\_\_\_

# Teaching Transparency Master Chemistry Answers

**G Orfield**



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