

Name:

ATOMIC STRUCTURE

1 Use the words in the box to answer the following questions.

electron, neutron, nucleus, proton, shells



- (a) The central part of an atom is called the
- (b) The has a negative charge
- (c) The has no charge
- (d) The nucleus contains the and the
- (e) Electrons are found in
- (f) The and the have the same mass.

2 An atom contains 32 protons and 48 neutrons.

- (a) What is its atomic number?
- (b) How many electrons does it have?
- (c) What is the mass of the atom?

3 Complete the table to show the missing numbers of particles in the atoms A - E.

| Atom | Protons | Neutrons | Electrons | Mass |
|------|---------|----------|-----------|------|
| A | 20 | | | 40 |
| B | | | 21 | 45 |
| C | | 24 | | 44 |
| D | 19 | 21 | | |
| E | | | 22 | 45 |

Which two atoms are isotopes of the same element? and

4 Complete the table to show the missing information:

| Atom | Atomic number | Electron arrangement | Group in the Periodic Table | Period in the Periodic Table |
|------|---------------|----------------------|-----------------------------|------------------------------|
| P | | 2, 8, 7 | | |
| Q | 11 | | | |
| R | | | 4 | 2 |
| S | | | 3 | 3 |

Atomic Structure Test Questions

S Ashworth



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