

Name Mel

Locker # _____

Bonding & Molecular Structure**LEWIS STRUCTURES**

Indicate the total # of VALENCE electrons for each molecule (VE's). Calculate the number of bonded electrons from the skeleton (BE's). Calculate the Remaining Electrons (RE's = VE's - BE's) and distribute the remaining electrons onto the skeletal structure. Finally, satisfy the octets by creating double and triple bonds. Your final structures should be given below and your scratch paper should clearly show your work.

| | | | |
|---|--|---|--|
| F_2 VE's = <u>14</u> BE's = <u>2</u> RE's = <u>12</u> $:\ddot{F}-\ddot{F}:$ Shape: <u>linear</u> | H_2 VE's = <u>2</u> BE's = <u>2</u> RE's = <u>0</u> $H-H$ Shape: <u>linear</u> | HF VE's = <u>8</u> BE's = <u>2</u> RE's = <u>6</u> $H-\ddot{F}:$ Shape: <u>linear</u> | NH_3 VE's = <u>8</u> BE's = <u>6</u> RE's = <u>2</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{N}}-H$ Shape: <u>trigonal pyramidal</u> |
| CH_4 VE's = <u>8</u> BE's = <u>8</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape: <u>tetrahedral</u> | C_2H_6 VE's = <u>14</u> BE's = <u>14</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape: <u>tetrahedral (both)</u> | AlH_3 VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $H-Al-H$ Shape: <u>trigonal planar</u> | BH_3 VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{B}}-H$ Shape: <u>trigonal planar</u> |
| C_2H_4 VE's = <u>12</u> BE's = <u>12</u> RE's = <u>0</u> $H-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}=\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-H$ Shape around one C: <u>trigonal planar</u> | C_2F_4 VE's = <u>36</u> BE's = <u>12</u> RE's = <u>24</u> $:\ddot{F}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}=\overset{\cdot\cdot}{\underset{\cdot\cdot}{C}}-\ddot{F}:$ Shape around one C: <u>trigonal planar</u> | CO VE's = <u>10</u> BE's = <u>6</u> RE's = <u>4</u> $\overset{\cdot\cdot}{C}\equiv\overset{\cdot\cdot}{O}$ Shape: <u>linear</u> | O_2 VE's = <u>12</u> BE's = <u>4</u> RE's = <u>8</u> $:\overset{\cdot\cdot}{O}=\overset{\cdot\cdot}{O}:$ Shape: <u>linear</u> |

Lewis Dot Structure Work Sheet Answer Key

Jin-Ying Zhang



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Table of Contents Lewis Dot Structure Work Sheet Answer Key

1. Understanding the eBook Lewis Dot Structure Work Sheet Answer Key
 - The Rise of Digital Reading Lewis Dot Structure Work Sheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Lewis Dot Structure Work Sheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lewis Dot Structure Work Sheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lewis Dot Structure Work Sheet Answer Key
 - Personalized Recommendations
 - Lewis Dot Structure Work Sheet Answer Key User Reviews and Ratings
 - Lewis Dot Structure Work Sheet Answer Key and Bestseller Lists

5. Accessing Lewis Dot Structure Work Sheet Answer Key Free and Paid eBooks
 - Lewis Dot Structure Work Sheet Answer Key Public Domain eBooks
 - Lewis Dot Structure Work Sheet Answer Key eBook Subscription Services
 - Lewis Dot Structure Work Sheet Answer Key Budget-Friendly Options
6. Navigating Lewis Dot Structure Work Sheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Lewis Dot Structure Work Sheet Answer Key Compatibility with Devices
 - Lewis Dot Structure Work Sheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lewis Dot Structure Work Sheet Answer Key
 - Highlighting and Note-Taking Lewis Dot Structure Work Sheet Answer Key
 - Interactive Elements Lewis Dot Structure Work Sheet Answer Key
8. Staying Engaged with Lewis Dot Structure Work Sheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lewis Dot Structure Work Sheet Answer Key
9. Balancing eBooks and Physical Books Lewis Dot Structure Work Sheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lewis Dot Structure Work Sheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Lewis Dot Structure Work Sheet Answer Key
 - Setting Reading Goals Lewis Dot Structure Work Sheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lewis Dot Structure Work Sheet Answer Key
 - Fact-Checking eBook Content of Lewis Dot Structure Work Sheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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