

COMPUTER STUDIES

FORM TWO END TERM 2 EXAMINATION-2022

1. Name **three** major components of a computer system. (3 marks)

.....
.....
.....

2. John saw the following symbols at the back of his system unit. What do the symbols represent?

(2 marks)



A



B

A

B

3. List down any **four** advantages of using computers as tools of problem solving. (2marks)

.....
.....
.....

4. Since the invention of the first generation of digital computers, much advancement has been realized in the sector of information and technology. Explain **two** characteristics which have been improved from the first generation to the modern computers. (4 marks)

.....
.....
.....

Computer Studies Questions And Answers Zimsec

Peipei Pang



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Table of Contents Computer Studies Questions And Answers Zimsec

1. Understanding the eBook Computer Studies Questions And Answers Zimsec
 - The Rise of Digital Reading Computer Studies Questions And Answers Zimsec
 - Advantages of eBooks Over Traditional Books
2. Identifying Computer Studies Questions And Answers Zimsec
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Computer Studies Questions And Answers Zimsec
 - User-Friendly Interface
4. Exploring eBook Recommendations from Computer Studies Questions And Answers Zimsec
 - Personalized Recommendations
 - Computer Studies Questions And Answers Zimsec User Reviews and Ratings
 - Computer Studies Questions And Answers Zimsec and Bestseller Lists
5. Accessing Computer Studies Questions And Answers Zimsec Free and Paid eBooks
 - Computer Studies Questions And Answers Zimsec Public Domain eBooks
 - Computer Studies Questions And Answers Zimsec eBook Subscription Services
 - Computer Studies Questions And Answers Zimsec Budget-Friendly Options
6. Navigating Computer Studies Questions And Answers Zimsec eBook Formats

- ePub, PDF, MOBI, and More
 - Computer Studies Questions And Answers Zimsec Compatibility with Devices
 - Computer Studies Questions And Answers Zimsec Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Computer Studies Questions And Answers Zimsec
 - Highlighting and Note-Taking Computer Studies Questions And Answers Zimsec
 - Interactive Elements Computer Studies Questions And Answers Zimsec
 8. Staying Engaged with Computer Studies Questions And Answers Zimsec
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Computer Studies Questions And Answers Zimsec
 9. Balancing eBooks and Physical Books Computer Studies Questions And Answers Zimsec
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Computer Studies Questions And Answers Zimsec
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Computer Studies Questions And Answers Zimsec
 - Setting Reading Goals Computer Studies Questions And Answers Zimsec
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Computer Studies Questions And Answers Zimsec
 - Fact-Checking eBook Content of Computer Studies Questions And Answers Zimsec
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect to zero energy at zero displacement from the spring s unstretched length $x = 0$ is $u = \frac{1}{2} kx^2$ figure 8 5 2 8 5 2 a glider between springs on an air track is an example of a horizontal mass spring system

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δh is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web pertanyaan serupa iklan perhatikan diagram energi potensial berikut pada diagram di atas energi aktivasi untuk reaksi baliknya ditandai dengan

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web mar 17 2016 3 what is the activation energy of a reaction and how is this energy related to the activated complex of the reaction E_a is the minimum amount of energy for a reaction to occur it is the amount of energy required to create an activated complex 4 what happens when a catalyst is used in a reaction a catalyst changes the reaction

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate Δh for the reaction and label it with the correct value see attached image

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need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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