

## 12

## STOICHIOMETRY

## Practice Problems

In your notebook, solve the following problems.

## SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups rasp and blueberries, and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate,  $\text{KClO}_3$ . Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from  $3.8 \times 10^{22}$  molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

## SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(III) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 15.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. (Write a balanced equation first.)

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of nitrogen dioxide. Assume all gases are at the same conditions of temperature and pressure.



# Chapter 12 Assessment Stoichiometry Answer Key

**LL Leslie**



**Chapter 12 Assessment Stoichiometry Answer Key:**

## Adopting the Track of Appearance: An Mental Symphony within **Chapter 12 Assessment Stoichiometry Answer Key**

In some sort of consumed by screens and the ceaseless chatter of quick connection, the melodic elegance and mental symphony produced by the published word frequently disappear into the back ground, eclipsed by the persistent sound and distractions that permeate our lives. But, located within the pages of **Chapter 12 Assessment Stoichiometry Answer Key** a marvelous fictional prize filled with natural emotions, lies an immersive symphony waiting to be embraced. Crafted by a wonderful composer of language, this captivating masterpiece conducts visitors on an emotional trip, well unraveling the hidden melodies and profound influence resonating within each carefully constructed phrase. Within the depths of this touching analysis, we can discover the book is central harmonies, analyze its enthralling writing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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